

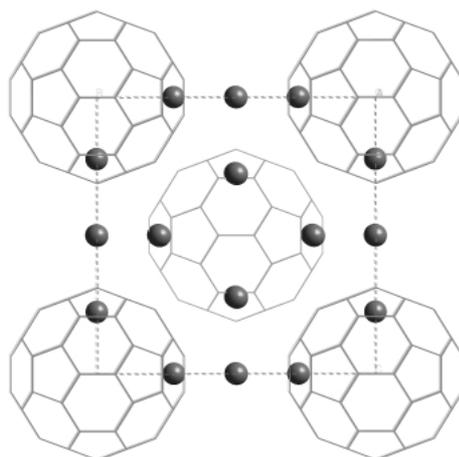


	<b>Experiment title:</b> Structural Studies of Superconducting Alkali and Alkaline Earth Fullerides	<b>Experiment number:</b> CH-731
<b>Beamlines:</b> BM16/BM1	<b>Date of experiment:</b> from: 17/11/99 to 22/11/99 and 26/1/00 to: 29/1/00	<b>Date of report:</b> 19/8/00
<b>Shifts:</b> 15/9	<b>Local contact(s):</b> A N Fitch / K Knudsen	<i>Received at ESRF:</i>
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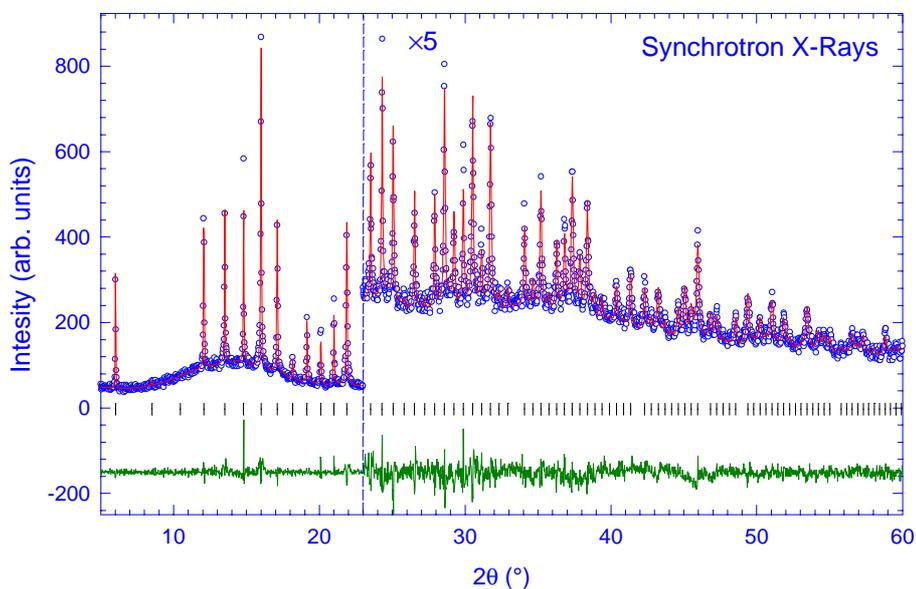
**Report:**

Abstract of publication resulting from this proposal: *Chem. Mater.* **12**, Sept. 2000, in press.

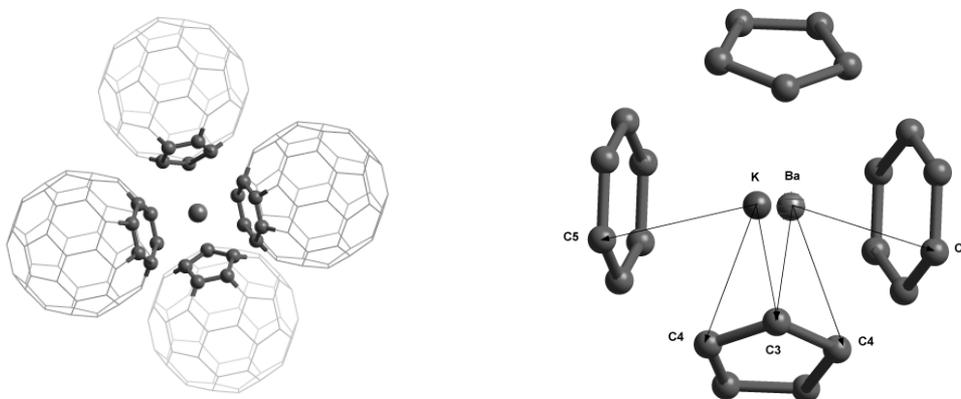
The crystal structure of the superconducting  $K_3Ba_3C_{60}$  fulleride ( $T_c = 5.4$  K) has been studied by synchrotron X-ray and neutron powder diffraction between 10 and 295 K. It is body-centered cubic (*bcc*) at all temperatures with an essentially half full  $t_{1g}$ -derived band. Close contacts between  $Ba^{2+}$  and  $K^+$  ions and neighboring  $C_{60}$  units imply a strong orbital hybridization, which leads to broadening of the conduction band. We find that the  $Ba^{2+}$  and  $K^+$  cations are disordered in the same distorted tetrahedral interstitial sites,  $(0, \frac{1}{2}, \frac{1}{4} + \delta)$  but they are displaced by a different distance,  $\delta$  from the center of the site. The resulting local distortions may be responsible for the observed complex relationship between  $T_c$  and cubic lattice parameter in  $A_3Ba_3C_{60}$  ( $A = Na, K, Rb, Cs$ ) fullerides.



**Fig. 1** Schematic view of the *bcc* structure of  $\text{K}_3\text{Ba}_3\text{C}_{60}$  on the (001) plane. Shaded circles show the interstitial sites, which are randomly occupied by alkali and alkaline-earth metals.



**Fig. 2** Final observed (points) and calculated (solid line) synchrotron X-ray ( $\lambda = 0.83502 \text{ \AA}$ ) powder diffraction profile for  $\text{K}_3\text{Ba}_3\text{C}_{60}$  at 10 K. The lower line shows the difference profiles and the ticks mark the position of the Bragg reflections. The profile has been expanded for clarity by a factor of 5 at Bragg angles larger than  $23^\circ$ .



**Fig. 3** Perspective view of the coordination of the  $\text{Ba}^{2+}$  and  $\text{K}^+$  cations in the *bcc* structure of  $\text{K}_3\text{Ba}_3\text{C}_{60}$ .