



**Experiment title: XAFS study of Ti-M binary mixed oxides used for photo-catalytic elimination of pollutants**

**Experiment number:**  
ME-605

**Beamline:**

**Date of experiment:**  
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**Shifts:**

18

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## Report:

Photocatalysis is a novel field of great interest for catalytic research with a significant potential to solve environmental problems. The use of anatase-TiO<sub>2</sub> as the base system for photocatalytic processes and the need to further optimize this reference catalyst are well known facts. In particular, its technological application seems limited by several factors among which the most restrictive one is the need of using an ultraviolet (UV) excitation source. The efficient use of solar light or, in other words, of the visible region of the spectrum, may then appear as an appealing challenge for developing the future generation of photocatalytic materials. Recent advances in this field make use of anatase-like Ti-M mixed oxide; the presence of some heteroatoms, like W, into the anatase structure yields the required decrease of the anatase band gap energy and the enhancement of the visible absorption power. The XAFS characterization of these complex oxides is thus necessary in order to interpret their chemico-physical properties.

XAFS experiments were carried at the beam line BM29 with a Si(111) monochromator and the station He cryostat (EXAFS measurements). Series of 5 to 9 Ti-M mixed oxides with growing quantities of

the heteroatom were studied for  $M = W$  and  $Zr$ . Both heteroatoms induce beneficial effects in catalytic elimination of organic pollutants with respect to the bare anatase- $TiO_2$  reference but  $Zr$  makes this only under UV-light excitation while  $W$  is able to do it under both UV- and visible-light excitation. Laboratory characterization show that all samples are mixed oxides with anatase-type structure. XANES spectra (Fig. 1) at the Ti-K edge further support the anatase-type crystalline structure of all mixed oxides while EXAFS measurements indicates little differences in the first Ti-O coordination shell with respect to the monocationic  $TiO_2$  reference.

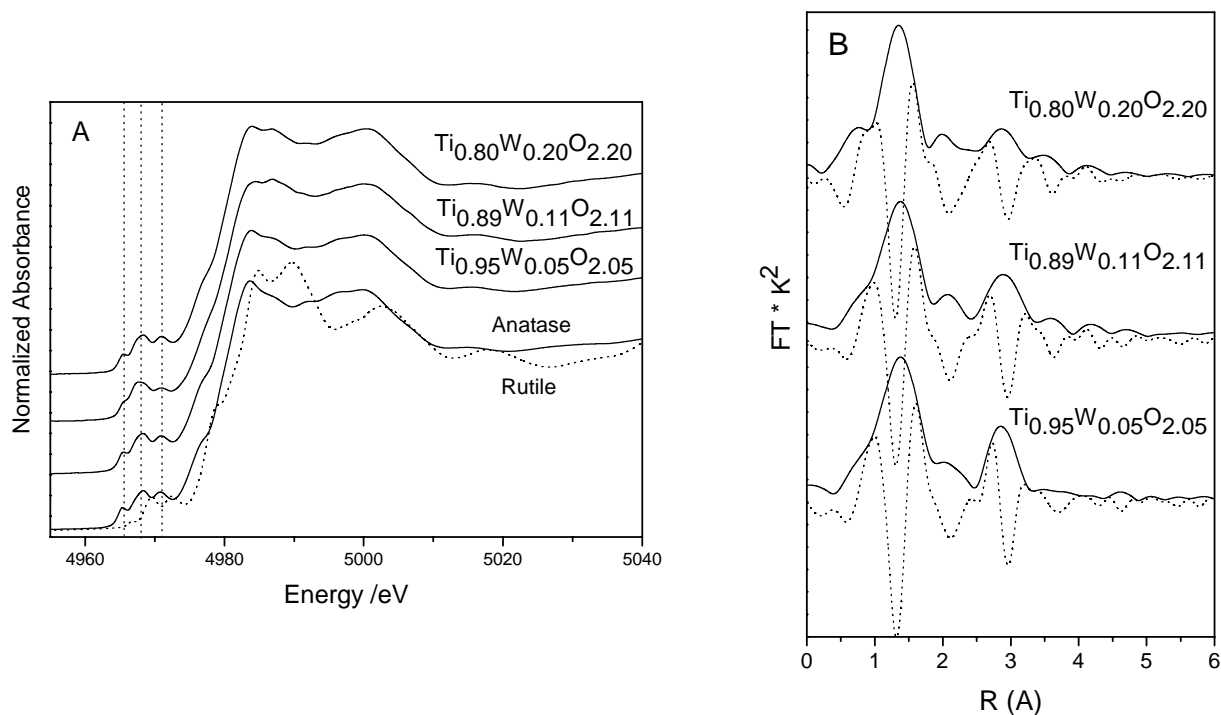


Fig.1. A) Ti K-edge XANES spectra and B) W  $L_{III}$ -edge Fourier Transform of EXAFS spectra for Ti-W mixed oxide samples.

In the case of Ti-W mixed oxides, analysis of the W  $L_I$ -edge XANES spectra indicates that W occupies low symmetry positions, displaying strong local disorder with respect to cationic positions characteristic of the anatase structure. Analysis of the W  $L_{III}$ -edge EXAFS spectra (Fig. 1B) suggests that W has a similar oxygen first coordination and exclusively Ti as second neighbor. Presence of W in the material produces cationic vacancies at first (cationic) neighbor distance in order to achieve charge neutrality. Additionally, above a 15 at. W %, the EXAFS analysis shows the existence of local order with a number a number of vacancies around W centers growing from 1/2 below/above this point. For Ti-Zr mixed oxides, the XAFS study shows that Zr also occupies strongly distorted substitutional positions of the anatase structure, having a rather low symmetry with respect to those typical of the parent  $TiO_2$  material. This occurs with a concomitant modification of *i*) the  $sp(d)$  electronic structure of Zr (W) in a modest way, as measured by XANES, and *ii*) of the conduction band of the anatase-based nano-materials. These XANES/EXAFS results, published in two journals *Journal of Physical Chemistry B* [1] and *Chemistry of Materials* [2], indicate that substitutional substitution of Ti on anatase occurs with a characteristic local order and, together with

optical measurements, show the modification of the anatase conduction band and band gap. They thus are expected to give adequate input data for theoretical calculations which may fully interpret the band gap properties of these Ti-based mixed oxide materials.

Assistance of BM29 staff (Dra. S. Ramos) during this experiment is gratefully acknowledged.

## **References**

- 1.- M. Fernández-García, A. Fuerte, A. Martínez-Arias, J.C. Conesa, *J. Phys. Chem. B* 109 (2005) 6075-6083.
- 2.- M.D. Hernández-Alonso, B. Bachiller-Baeza, J.M. Coronado, M. Fernández-García, J. Soria, *Chem. Mater.* 19 (2007) 4283-4291.