

  ROBL-CRG	<b>Experiment title:</b> Investigation of $\text{Th}_x\text{Zr}_{(1-x)}\text{SiO}_4$ and $\text{U}_x\text{Zr}_{(1-x)}\text{SiO}_4$ Solid Solutions: colloid formation and solid state characteristics	<b>Experiment number:</b> 20-01-745
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## Report:

Commercially used fuel rods use zircalloy claddings of with high-purity zirconium as a first barrier. Zirconium will therefore be present in a deep geological repository for spent nuclear fuel. On the other hand zircon,  $\text{ZrSiO}_4$ , has been under investigation as a potential waste matrix for high-level radioactive wastes. Additionally, the formation of  $\text{U}_x\text{Zr}_{(1-x)}\text{SiO}_4$  in the so called Chernobyl lavas has been postulated [1]. While data on Pu doped  $\text{ZrSiO}_4$  exists [2], data on other  $\text{ZrSiO}_4 - \text{An(IV)SiO}_4$  solid solutions is scarce. Examining the mixing properties of the zircon-type orthosilicates such as  $\text{USiO}_4$ ,  $\text{ThSiO}_4$ ,  $\text{ZrSiO}_4$  and  $\text{HfSiO}_4$  can help to understand tendencies of thermodynamic properties in these systems and to deduce values, which are otherwise hard to obtain.

Complete solid solutions between  $\text{USiO}_4$  and  $\text{ThSiO}_4$  have been shown to be possible [3-5] and lattice parameters in the  $\text{U}_x\text{Th}_{(1-x)}\text{SiO}_4$  system exhibit behavior according to Vegard's Law.  $\text{Zr}_x\text{Hf}_{(1-x)}\text{SiO}_4$  solid solutions have also been described in detail recently [6] and exhibit a negative excess volume of mixing, indicator for non-ideal mixing behavior. To the best of our knowledge neither has the successful synthesis of  $\text{U}_x\text{Zr}_{(1-x)}\text{SiO}_4$  solid solutions nor that of

$\text{Th}_x\text{Zr}_{(1-x)}\text{SiO}_4$  been reported so far. First principle calculations by Ferriss et al. [7] suggest that there is a miscibility gap between  $\text{USiO}_4$  and  $\text{ZrSiO}_4$  and  $\text{ThSiO}_4$  and  $\text{ZrSiO}_4$  (calculated for 500 K and 1000 K). For the  $\text{HfSiO}_4$  and  $\text{ZrSiO}_4$  system their paper proposes near ideal mixing with a weak tendency for phase separation. The latter was later confirmed by Cota et al [6].

Stoichiometric solid  $\text{Th}_x\text{Zr}_{(1-x)}\text{SiO}_4$  and  $\text{U}_x\text{Zr}_{(1-x)}\text{SiO}_4$  samples were prepared from sol-gel synthesis through a hydrothermal procedure with the aim to obtain well defined material for XRD characterization and subsequent experiments. The  $\text{Th}_x\text{Zr}_{(1-x)}\text{SiO}_4$  ( $x=0.1, 0.2, \dots, 0.9$ ) solid solution series as well as  $\text{ZrSiO}_4$  was successfully synthesized and characterized prior to measurement by XRD, IR, and Raman. The aim of the present study is to investigate  $\text{U}_x\text{Zr}_{(1-x)}\text{SiO}_4$  and  $\text{Th}_x\text{Zr}_{(1-x)}\text{SiO}_4$  solid solutions regarding their mixing behavior. Refinement of the lattice parameters showed that the  $\text{Th}_x\text{Zr}_{(1-x)}\text{SiO}_4$  solid solutions follow Vegard's rule but contain a positive deviation in excess volume of mixing. It is notable, that especially  $a = b$  lattice parameters contribute to this effect, while the  $c$  lattice parameter behaves nearly ideal. Spectroscopic data from IR measurements show that the  $\nu_2$  and  $\nu_4$ -modes, associated to the  $\text{SiO}_4^{4-}$  tetrahedron shift towards smaller wavenumbers with increasing mole fraction  $x$ . The Raman study of these solid solutions shows the same effect for most modes: a linear decrease of the wavenumber with increasing  $x$ . EXAFS was used to investigate the local structure. The near-order structure in  $\text{Th}_x\text{Zr}_{(1-x)}\text{SiO}_4$  was investigated by using the Zr K-edge and Th  $L_3$ -edge. The Th  $L_3$  edge was used to determine Th-Zr and Th-Th distances and the Zr K edge was used to determine Zr-Th and Zr-Zr distances. XANES edge-step was used to determine the mole fraction  $x$ . The obtained values are confirmed by SEM-EDS analysis. It appears that the  $\text{ThSiO}_4 - \text{ZrSiO}_4$  system forms a complete solid solution, which show deviations from Vegards law. Results received from bulk material show significant structural differences related to colloidal samples [8].

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