



**Experiment title:**

**Cation location and its environment  
in Rb-exchanged zeolites.**

**Experiment  
number:**

**CH-107**

**Beamline: Date of Experiment:**

**Date of Report:**

**D8 GILDA** from: November 01 to: November 04, 1995

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**Shifts: Local contact(s):**

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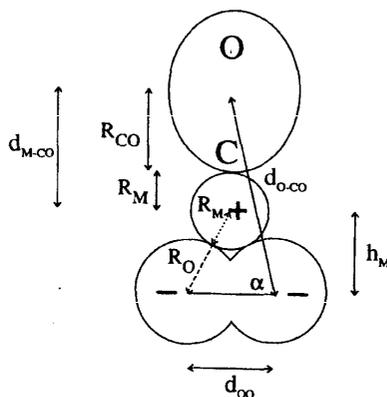
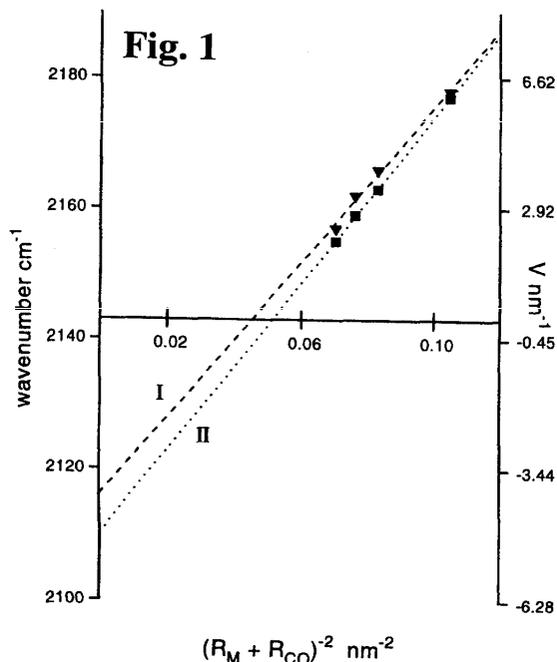
## **Report:**

In the last few years our group has performed an ongoing research program in the characterization of alkali-metal exchanged zeolites, mainly using infrared (IR) spectroscopy of adsorbed molecules [1-4]. In particular, a linear blue shift of the C-O stretching frequency of  $M^{+..}CO$  adducts in zeolites ( $M = Cs, Rb, K, Na, Li$ ), see Fig. 1, was found as a function of  $(RM + RCO)^2$ , being  $RM$  the cation radius and  $RCO$  the molecule radius. This trend enable us to affirm that the interaction is mainly of electrostatic nature and to infer the geometry of the  $M^{+..}CO$  adsorption site, see Fig.2. We have thus performed XAFS measurements on Rb-ZSM-5 and Rb-mordenite to verify our previous findings. The complexity of the experiment is remarkable because: **i**) a crytical outgassing procedure at 700 K must be followed to assure the evacuation of undesired adsorbed molecules; **ii**) after the dosage of a desired ammount of CO (or  $N_2$ ), the obtained pure atmosphere must be preserved during the XAFS data acquisition (4 h for sample cooling, cell location and 3 spectra collection); **iii**) being the  $Rb^{+..}CO$  interaction very weak, adducts can not be formed at RT and measurements at 77 K are required. To fulfill those requests, as planned in the proposal, we have realized an improved version of our EXAFS cell previously used at ADONE and LURE to measure zeolites at RT [5]. This new cell has operated for the first time at GILDA D8 beamline of ESRF during experiment CH- 107, where we have successfully observed  $Rb^{+..}CO$  and  $Rb^{+..}N_2$  adducts on ZSM-5 and mordenite. A detailed description of the samples treatments procedure and all technical details on the cell are reported elsewhere [6].

XAFS experiments have been carried out in transmission mode using a dynamical sagittal focused Si(3 11) monochromator; energy calibration has been performed with Kr gas. In the XANES part of the spectra, a detailed analysis of the white line area [7,8] assure that we are dealing, in vacuum conditions, with  $Rb^{+}$  species, giving so a further proof of the quality of the prepared samples.

Moreover, being the edge position and the white line area nearly unaffected upon CO and N<sub>2</sub> dosages, XANES data strongly confirm that the charge of Rb still remains +1 and that both Rb<sup>+</sup>...CO and Rb<sup>+</sup>...N<sub>2</sub> interactions are of electrostatic nature, in fully agreement with our previous findings [1-4]. EXAFS analysis is complex, because: **i**) Rb ions are surrounded (see Fig.2) by only two oxygens atoms, which have a weak backscattering power, so giving rise to a weak EXAFS signals; **ii**) Rb is affected by double electron excitation manifesting in a sharp peak at about 6. 1A<sup>-1</sup> [9- 10]; **iii**) the use of analysis program able to work in the framework of multiple scattering (MS) theory [11 ] is necessary for the spectra taken in presence of adsorbates, because for both Rb<sup>+</sup>...CO and Rb<sup>+</sup>...N<sub>2</sub> adducts a linear adsorption geometry is expected [1-4], see Fig.2, so enhancing the MS contributions. EXAFS analysis is still in progress at the moment and, from a fully qualitative ground, we can now only state that an increment of the Rb coordination number has been observed after CO and N<sub>2</sub> dosage.

We have also obtained 6 shifts for performing powder X-ray diffraction measurement on the same samples at ESRF D 16 beamline: experiments will be performed in the next weeks. We are indebted with all the GILDA D8 staff and in particular with our local contact F. D'Acapito and with the ESRF chemical lab. staff which have enabled us to operate in ideal conditions.



**Fig. 2**

## References

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