



Experiment title: EXAFS and high resolution XANES study of the location, aggregation and oxidative state of iron species in Fe-silicalite, Fe-Al-silicalite and Fe exchanged ZSM-5 and Y zeolites		Experiment number: CH-695
Beamline: BM8 GILDA	Date of experiment: from: 05/09/1999 to: 11/09/1999	Date of report: 31/08/2001
Shifts: 15	Local contact(s): Francesco D'Acapito	<i>Received at ESRF:</i>

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Report:

In previous works, reporting EXAFS/XANES data collected at LURE supported by IR, UV-Vis and EPR experiments, we have demonstrated that template removal at increasing temperature results in the progressive migration of Fe from framework into extraframework species [1-3].

The present study was aimed to obtain a detailed, high resolution, XANES picture to the determine both oxidation state and of local symmetry of Fe species upon template removal and interaction with N₂O [4-6]. Among the measured model compound, we mention: FePO₄ (tetrahedral Fe³⁺); α -Fe₂O₃, (highly distorted octahedral Fe³⁺); and Fe(acac)₃ (quite perfect octahedral Fe³⁺); FeCp₂ (Fe²⁺ in D_{5d} symmetry). Edge position, white line intensity (Fig. 1a), pre-edge intensity and position (Fig. 2a) well reflects the oxidation state and the local symmetry of iron species in the model compounds. The A₁ \rightarrow T₂ pre-edge peak of Fe-silicalite with template (Fig. 2b) is even higher than that of FePO₄ (Fig. 2a), indicating that Fe-silicalite with template is a better model than FePO₄ for Fe³⁺ in T_d symmetry. Fig. 1b and 2b also reports the effect of template burning, at increasing temperature on the XANES spectra. We note: (i) red shift of the edge (7123.6 \rightarrow 7122.4 \rightarrow 7120.6 eV); (ii) white line decrement (1.52 \rightarrow 1.31 \rightarrow 1.23); (iii) decrement (0.205 \rightarrow 0.142 \rightarrow 0.128) and broadening of the pre-edge peak; (iv) appearance of a new component at 7111.8 eV. Points (i) and (iv) highlight the reductive nature of the framework dislodgment process: migration to extraframework position is accompanied by Fe³⁺ \rightarrow Fe²⁺ reduction; IR and EPR data support this important finding [4-6]. Point (ii) prove the high coordinative unsaturation of extraframework species as supported by the low Fe coordination number(EXAFS) and by the formation of Fe³⁺(NO)₃ adducts (IR). Point (iii) simply means that we lost the initial T_d symmetry. Figs 1c and 2c prove that N₂O is able to re-oxidize extraframework Fe²⁺ species, as supported by IR and EPR [4-6].

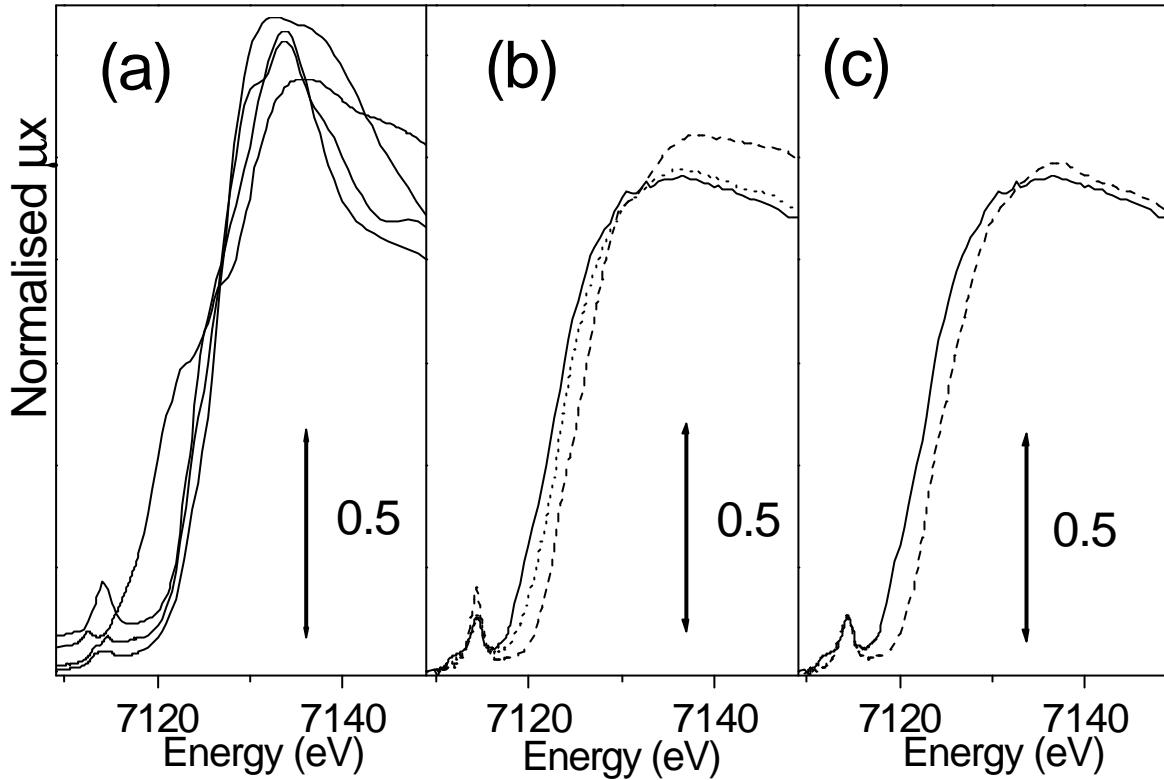


Figure 1 Part (a) XANES spectra of model compounds: from top to bottom FePO_4 , FeCp_2 , Fe_2O_3 , $\text{Fe}(\text{acac})_3$ (Cp = cyclopentadienyl, acac = Acetylacetone). The four spectra have been vertically shifted for clarity. Part (b) XANES spectra of Fe-Silicalite with template (dashed line), outgassed at 773 K and 973 K (dotted and full line respectively). Part (c) XANES spectra of Fe-silicalite outgassed at 973 K (full line, same curve as in part b, here reported again for direct comparison) and successively oxidised in N_2O at 523 K (dashed line).

References

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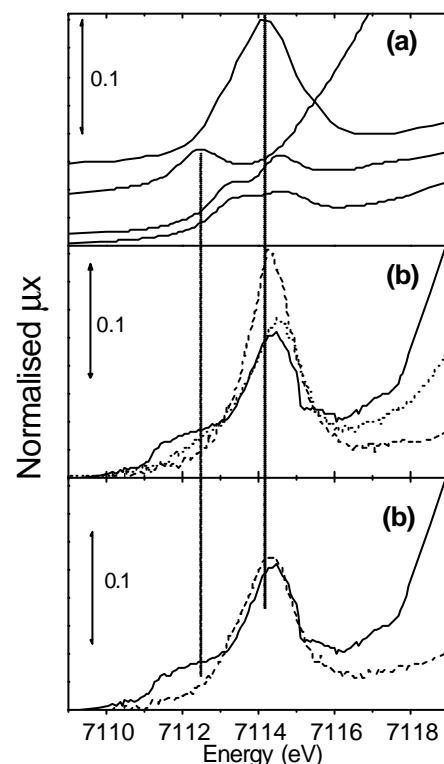


Fig.2 As Fig.1: zoom .