



Experiment title:
Environmental friendly pre-treatments for aluminium alloys: a XAFS study in Ce-doped silane films

Experiment number:
MA-238

Beamline: BM-29	Date of experiment: from: 20 Sept. 2006 to: 22 Sept. 2006	Date of report: June 2007
Shifts: 2	Local contact(s): Dr. Sakura Pascarelli	<i>Received at ESRF:</i>

Names and affiliations of applicants (* indicates experimentalists):

CARMEZIM, Maria-João, Politécnico de Setúbal, ESTSetubal, Mechanical Engineering Dept, Campus IPS, 2910 Setúbal, Portugal

FIGUEIREDO, Maria-Ondina, CENIMAT, New Univ. Lisbon, 2829-516 Caparica, Portugal

CABRAL, Ana, ISQ, Production Technologies, Tagus Park, 2740-120 Oeiras, & ISEL, Chem. Eng. Dep. 1959-007 Lisboa

Report:

EXAFS experiments on *bis-silane films doped with cerium* were done to investigate the structure around cerium ions, in this environment-friendly pre-treatment for aluminum alloys.

Several spectra were also collected from prototypes selected to model cerium speciations (valence plus coordination environment): synthetic CeO_2 (a cubic oxide with fluorite-type crystal structure and regular cubic coordination of Ce^{4+}), monazite (La-rich natural CePO_4 with a less regular environment of Ce^{3+} ions), octa-hydrated Ce^{4+} sulphate $\text{Ce}(\text{SO}_4)_2 \cdot 8\text{H}_2\text{O}$, and hexa-hydrated Ce^{3+} nitrate $\text{Ce}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$.

The experiments were performed in transmission mode at the Ce K-edge (40443 eV) using a Si (311) monochromater, with the collaboration of the beamline scientist Dr. Sakura Pascarelli.

Data from synthetic ceria (CeO_2) fitted in *R*-space by taking into account only single scattering contributions due to the first four shells of atomic neighbours and using the FEFF code is presented in the Figure 1a. The inverse fast Fourier transforms of experimental and theoretical EXAFS (Figure 1b) display a good agreement. Further work is being carried out on data XAFS collected from Ce hydrated salts (sulphate and nitrate) and from monazite, an important natural Ce-carrier. Results of these experiment have already been presented [1].

[1] - M. O. Figueiredo, M. J. Carmezim, A. M. Cabral, M. G. S. Ferreira "A XAFS STUDY AT CERIUM K-EDGE IN COMPOUNDS FIGURING OUT VARIOUS *Ce* -SPECIATIONS AND COORDINATION ENVIRONMENTS", IV International Materials Symposium, MATERIAIS 2007 – A Materials Science Forum, April, 1-4, Faculdade de Engenharia da Universidade do Porto, Oporto/Portugal.

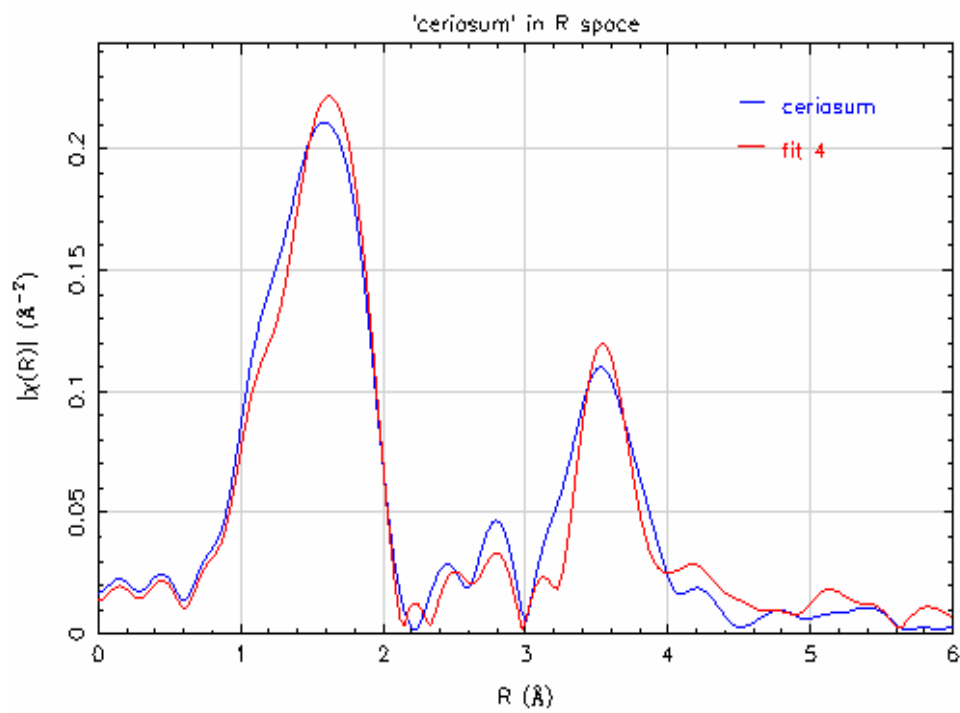


Figure 1(a)

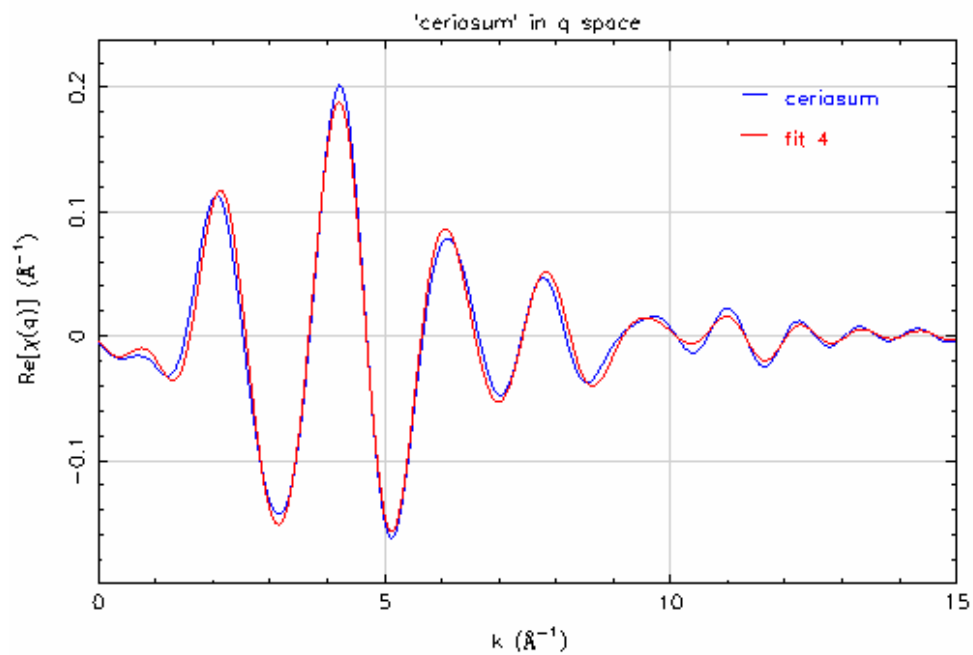


Figure 1(b)

A XAFS STUDY AT CERIUM K-EDGE IN COMPOUNDS FIGURING OUT VARIOUS Ce-SPECIATIONS AND COORDINATION ENVIRONMENTS

M. O. Figueiredo¹, M. J. Carmezim^{2,*}, A. M. Cabral³, M. G. S. Ferreira⁴

¹ CENIMAT, Mat. Sc. Dep., New Univ. Lisbon, 2829-516 Caparica, & INETI-IGM, Dep. Min. Res., Apt. 7586, 2721-866 Alfragide,

² Inst. Politécnico de Setúbal, ESTSetubal, Mechanical Engineering Dept, Campus IPS, 2910 Setúbal, Portugal

³ ISQ, Production Technologies, Tagus Park, 2740-120 Oeiras, & ISEL, Chem. Eng. Dep. 1959-007 Lisboa,

⁴ IST, Chem. Eng. Dep. 1049-001 Lisboa, & Univ. Aveiro, Dep. Ceramic and Glass Eng., 3810-193, Aveiro, Portugal

* Corresponding author: mjoao@est.ips.pt

Keywords: XAFS, Cerium compounds.

Abstract

The application fields of cerium compounds have enlarged in the last years. Cerium oxide is emerging as a hydrocarbon catalyst in self cleaning ovens and is now extensively used as a promoter in three-way catalysts (TWCs) for automotive emission control [1]. This oxide allows for selective absorption of ultraviolet light in glass and, owing to its nonstoichiometric behaviour once the oxidation state of Ce can be 3+ or 4+, CeO₂ also proved to be a good promoter for oxygen storage [2]. The cerium oxide is also used in diesel fuels for a more complete combustion to abate soot formation [3]. Moreover the magneto-optical signals of Ce-compounds have an outstanding figure of merit amongst other classes of magnetic materials [4].

Synchrotron radiation is nowadays a powerful means of studying X-ray absorption fine-structure spectroscopy (XAFS), either close to the absorption edge (XANES) or up to a hundred electron volts above the edge (EXAFS). The latter has revealed itself as an outstanding technique for structural characterization of catalytic systems [5].

As a preparatory approach to the characterization of corrosion behavior of bis-silane Ce-treated films over aluminum alloys [6], a XAFS study at the Ce K-edge in compounds with well-known crystal structure and chemical composition was undertaken at the ESRF[#] using the instrumental set-up of beamline BM29. The following prototypes were selected to model cerium speciations (valence plus coordination environment): synthetic CeO₂ (a cubic oxide with fluorite-type crystal structure and regular cubic coordination of Ce⁴⁺), monazite (La-rich natural CePO₄ with a less regular environment of Ce³⁺ ions), octa-hydrated Ce⁴⁺ sulphate Ce(SO₄)₂·8H₂O, and the hexa-hydrated Ce³⁺ nitrate Ce(NO₃)₃·6H₂O. A discussion is presented on the local structure parameters extracted from collected EXAFS spectra, along with prospective comments concerning the role of cerium in the pre-treatment of bis-silane films deposited onto alloy substrates.

References

- [1] Daniel M. Lyons *et al.*, Preparation of ordered mesoporous ceria with enhanced thermal Stability, *J. Mater. Chem.* 12: 1207–1212, 2002.
- [2] O. Adamopoulos *et al.*, The Characterization and Reactivity of Nanostructured Cerium-Copper-Oxide Composites for Environmental Catalysis, *Mat. Res. Soc. Symp. Proc.*: 676, 2001.
- [3] Zhang *et al.*, Cerium oxidation state in ceria nanoparticles studied with X-ray photoelectron spectroscopy and absorption near edge spectroscopy, *Surf. Sci.* 563: 74-82, 2004.
- [4] Pittini R.; Wachter P., Cerium compounds: The new generation magneto-optical Kerr rotators with unprecedented large figure of merit, *J. Magnetism & Magnetic Materials* 186: 306-312, 1998.
- [5] E. Fonda *et al.*, EXAFS analysis of the L₃ edge of Ce in CeO₂: effects of multi-electron excitation and final-state mixed valence. *J. Synchrotron Rad.* 6: 34-42, 1999.
- [6] A.M. Cabral *et al.*, Analytical characterization and corrosion behaviour of bis-[triethoxysilylpropyl] tetrasulphide pre-treated AA2024-T3, *Corrosion Sci.* 47: 869-881, 2005.

[#] The financial support of EU through the action Access to Research Infrastructures is acknowledged. Special thanks are due to the beam line Scientist at the ESRF, Dr. Sakura Pascarelli.