Fluorescence EXAFS data were collected on the dedicated chamber operative at the GILDA beamline. The monochromator was equipped with a Si(111) crystal pair and was run in dynamically focusing mode. The rejection of higher order harmonics was achieved by using a pair of Pt mirrors with an energy cutoff of 10 keV. X-ray fluorescence of the samples was measured by means of a thirteen elements Ge detector.

The samples were obtained according to the following protocol:

i) Sputtering in vacuum (10^{-2} mbar) of ZnO onto an Al₂O₃ single crystals (11-20), (0001) and (1-102) at Room Temperature (RT). The layer thickness was checked by X-ray reflectivity and resulted to be 150 Å. The chemical state of the layer was checked by X-ray diffraction and resulted to be metallic ZnO.

ii) Firing of the ZnO layer with the substrate. The treatments were carried out at 300°C for 20 minutes and 800°C for different amount of time in air.

For comparison purposes the EXAFS spectra in conventional transmission mode of ZnO and $ZnAl_2O_4$ powders were also collected.



Figura 1: Zn K-edge XANES spectra of ZnO onto Al_2O_3 (11-20) single crystal after different treatments. For comparison purposes the inset also shows the spectra of ZnO and ZnAl₂O₄ powders.



Figura 2: Zn K-edge XANES spectra of ZnO onto Al_2O_3 (0001) single crystal after different treatments. For comparison purposes the inset also shows the spectra of ZnO and ZnAl₂O₄ powders.



Figura 3: Zn K-edge XANES spectra of ZnO onto Al_2O_3 (0001) single crystal after different treatments. For comparison purposes the inset also shows the spectra of ZnO and ZnAl₂O₄ powders.

The most striking result that can be easily observed looking at figures 1 - 3 is that differently oriented Al₂O₃ single crystals show impressive different reactivity towards ZnO. In particular:

- 1) (11-20) orientation: at 800 °C the spinel is formed and the amount of spinel increases with time, the sample appearing totally reacted after 2 hours at 800 °C.
- 2) (0001) orientation: at 800 °C a chemical reaction takes place, the product of which is different from both the spinel and ZnO. At the moment the chemical nature of this product has been not yet identified.
- 3) (1-102) orientation: The behaviour of this orientation is strongly resembling of that previously described for the (0001) orientation.
- 4) In addition, linear dichroism measurements at the Zn K-edge for the samples treated at 300° C showed that the ZnO thin film is highly oriented when deposited onto Al₂0₃ (11-20) and (0001) surfaces (the orientation of the film is with the c-axis of the wurtzite crystal structure perpendicular to the surface), while onto the Al₂0₃ (1-102) surface a polycrystalline film is formed.