



	Experiment title: Gas-phase dimerisation of acetic acid via hydrogen bonding	Experiment number: CH-3898
Beamline: ID20	Date of experiment: from: 2 October 2013 to: 8 October 2013	Date of report: 13 October 2014
Shifts: 18	Local contact(s): Ali Al-Zein	<i>Received at ESRF:</i>
Names and affiliations of applicants (* indicates experimentalists): J. Inkinen*, S. Huotari*, Ch.J. Sahle*, J. Niskanen, K. Hämäläinen Department of Physics, University of Helsinki, P.O.B. 64, FIN-00014 University of Helsinki, Finland		

Report:

We performed an inelastic X-ray scattering (IXS) experiment to study the dimerization of acetic acid (AA, CH_3COOH) in solid, liquid, and gaseous phases. The aim was to elucidate the underlying mechanism of changes in the electronic structure due to the complex formation of a hydrogen bond (HB) in a simple molecule. The dimerisation of AA has been studied previously with various techniques [1-5], and there has been discussion on the presence of the dimers in different conditions.

The IXS spectra were obtained with the X-ray Raman Scattering spectrometer of the beamline ID20. The incident beam was monochromatized with a Si(111) pre-monochromator and a Si(311) channel-cut monochromator, and the spectra were recorded using the 72 bent Si(660) analyzer crystals and two dimensional detectors.

For the gas-phase measurements a novel sample chamber setup was constructed. The actual chamber was a quartz capillary, which was oriented parallel to the incident beam to maximize the probed sample volume and thus the scattering signal. The capillary was in turn connected by glueing to the rest of the setup, including a container for liquid acid, a pressure gauge, and a pressure regulator. After loading the AA liquid to the sample chamber setup, it was evacuated to remove contribution of air. The liquid container was heated using a resistive heating element, thus vaporizing some of the liquid and allowing control of the pressure of the gas. To control the temperature of the probed volume of the gas, the capillary was covered with a copper tube, which was heated by resistive elements.

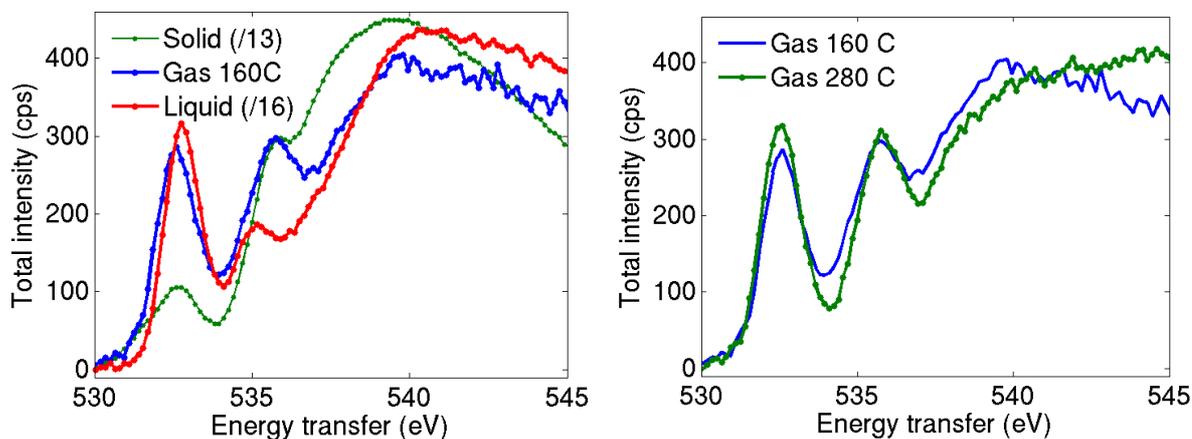


Figure 1: IXS spectra from oxygen K -edge of different phases of AA (left panel), and from gaseous AA at different temperatures (right panel).

For the liquid phase measurements the AA was loaded directly to the capillary. For the solid phase measurements a nitrogen flow from a cryostream sample cooler was used to lower the sample temperature to $-40\text{ }^{\circ}\text{C}$.

The preliminary results of the measurements from the oxygen K -edge of different phases of AA are shown in the left panel of Fig. 1. This comparison shows the sensitivity of the spectrum to the phase. The spectra from gaseous AA at different temperatures in the right panel show no clear signs of the dimer formation, contrary to what was expected based on the model of Ref. 5. The data analysis is in progress, which is focused to the examination of the effect of the H-bonding to the spectra. The publication based on this experiment is in course of preparation.

References:

- [1] K. Tabayashi et al. *Core-electron excitation and fragmentation processes of hydrogen bonded acetic-acid clusters in the oxygen K -edge region*, J. Electron. Spectrosc. Relat. Phenom. **184**, (2011), 134-139
- [2] D. Duflot et al. *Electronic excitation of gaseous acetic acid studied by K -shell electron energy loss spectroscopy and *ab initio* calculations*, Int. J. Mass spectrom.
- [3] J. J. Orlando et al. *Gas phase UV absorption spectra for peracetic acid, and for acetic acid monomers and dimers*, J. Photochem. Photobiol. A **157**, (2003) 161166
- [4] J. M. Briggs et al., *Monte Carlo simulations of liquid acetic acid and methyl acetate with the OPLS potential functions*, J. Phys. Chem., **95** (1991), 3315
- [5] A. Jónasson et al., *Vapor-liquid equilibria of systems containing acetic acid and gaseous components. Measurements and calculations by a cubic equation of state*, Fluid Phase Equilib., **152**, (1998), 67