ESRF	<b>Experiment title:</b> Structural and phase transitions of anodic alumina at high temperatures	<b>Experiment</b> <b>number</b> : MA-2696
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Names and affiliations of applicants (* indicates experimentalists): Kirill Napolskiy*, Ilya Roslyakov*, Eduard Levin* (Moscow State Univ.) Tatyana Kardash* (Novosibirsk State Univ.)		

Diffraction experiments were carried out on four series of anodic aluminas, prepared by anodization of high purity aluminium in different acid electrolytes (0.3 M  $H_2C_2O_4$ , 0.3 M  $H_2SO_4$  and 0.1 M  $H_3PO_4$ ). To check phase formation during annealing, a set of pre-annealed specimens was measured, and a number of *in situ* experiments were conducted. For comparison, total scattering data were also collected for several chemically synthesized aluminas of various polymorphs.

It turned out that, no matter of acid used, all as-prepared anodic aluminas are amorphous (Fig. 1, left panel). All observed distances are in good agreement with those reported in literature [1], and are, namely Al-O bond in AlO<sub>4</sub> tetrahedra (1.82 Å), O-O bond along AlO<sub>4</sub> tetrahedra (2.95 Å), Al-Al bond from corner-sharing tetrahedra (3.21 Å). Distances of 4.17 and 4.49 Å correspond, either to third coordination sphere, or to inclusion of foreign atoms, as sulfur or phosphorous, into the amorphous structure.

The structure of oxides after annealing slightly above crystallization temperature, as determined by preliminary TGA/DSC measurements, was examined (Fig. 1, right panel). On a local scale, all crystallized anodic oxides are spinel-type  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>. The difference in peaks intensities, starting from 10 Å, is caused by a different size of diffracting domains.



Fig. 1. PDFs of amorphous aluminas, prepared under different anodization conditions (left panel) and PDFs of the same aluminas after crystallization at 950 °C (right panel, top) compared to PDF of chemically synthesized γ-Al<sub>2</sub>O<sub>3</sub> (right panel, bottom). Anodic alumina prepared in phosphoric acid at 185 V is marked as AAP\_185V, in sulfuric acid at 25 V as AAS\_25V, in oxalic acid at 40 and 120 V as AAO\_40V and AAO\_120V, respectively.

Aluminas crystallization pathways were studied in a set of *in situ* experiments during isothermal annealing at 950 °C. An example for one of the samples, AAO\_120V, is given in Fig. 2. Despite of multiple possibilities of low-temperature polymorph intermediates formation [2], no such phases were observed – all aluminas crystallized into spinel-type structure.



Fig. 2. A set of PDFs taken in situ for AAO 120V during isothermal annealing at 950 °C.

The high-temperature *in situ* experiments in mirror furnace were cancelled due to extremely intense and non-uniform contribution from platinum capillary, which was much stronger than the signal from alumina samples and does not allowed for correct substraction of its contribution during patterns normalization.

Finally we would like to thank Carlotta Giacobbe for her excellent support during the experiment.

## **References:**

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2. G. Busca, Structural, surface, and catalytic properties of aluminas, in Advances in catalysis (F. Jentoft, Ed.), V. 57, Ch. 3, Amsterdam: Academic Press, 2014.