



ESRF

Experiment title:

Structural studies of heterofullerenes

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Report:

The recent synthesis and isolation in bulk quantities of the nitrogen-substituted fullerene solid,¹(C₅₉N)₂, and its subsequent intercalation with alkali metals to afford azafulleride salts² with stoichiometry A₆C₅₉N (A= K, Rb) opened the way to new opportunities in the quest for the synthesis of fullerene-based materials with novel structural, electronic and conducting properties. Unlike solid C₆₀, whose properties have been exhaustively studied in recent years, little is known at present about the physical properties of such condensed heterofullerene phases. As part of our present beam allocation on BMI, we performed X-ray diffraction measurements on a series of alkali metal intercalated (C₅₉N)₂ as a function of temperature. A preliminary account of our results is currently in press.³ In brief, A₆C₅₉N (A= K, Rb) adopt bcc structures, comprising of quasi-spherical monomeric (C₅₉N)⁶⁻ ions and are essentially isostructural with the corresponding salts of C₆₀, K₆C₆₀ and Rb₆C₆₀. On the other hand, Na₆C₅₉N adopts a hexagonal close packed (space group *P63/mmc*, a= 10.257(2) Å, c= 16.745(11) Å, c/a= 1.633) structure (Fig. 1), reflecting the smaller size of the Na⁺ cation, in the same way that the same effect leads to Na₆C₆₀ adopting a face centred cubic structure.

In addition, attempts to synthesise a series of Rb intercalates of (C₅₉N)₂ with varying Rb

content have resulted in weakly-doped paramagnetic hexagonal salts with stoichiometries $\text{Rb}_{2x}(\text{C}_{59}\text{N})_2$ ($x < 0.9$) (Fig. 1). The structural data are consistent with charge transfer from Rb to the heterofullerenes as the lattice dimensions show slight contraction upon intercalation ($a = 9.81 \text{ \AA}$, $c = 16.06 \text{ \AA}$ for $\text{Rb}_{1.8}(\text{C}_{59}\text{N})_2$).

References

[1] Hummelen, J.C. *et al.*, *Science* 1995,269, 1554.

[2] Prassides, K. *et al.*, *Science* 1996,271, 1833.

[3] Kordatos, K. *et al.*, in Proceedings of International Winterschool on Novel Nanostructures, Eds. H. Kuzmany *et al.*, 1997, in press.

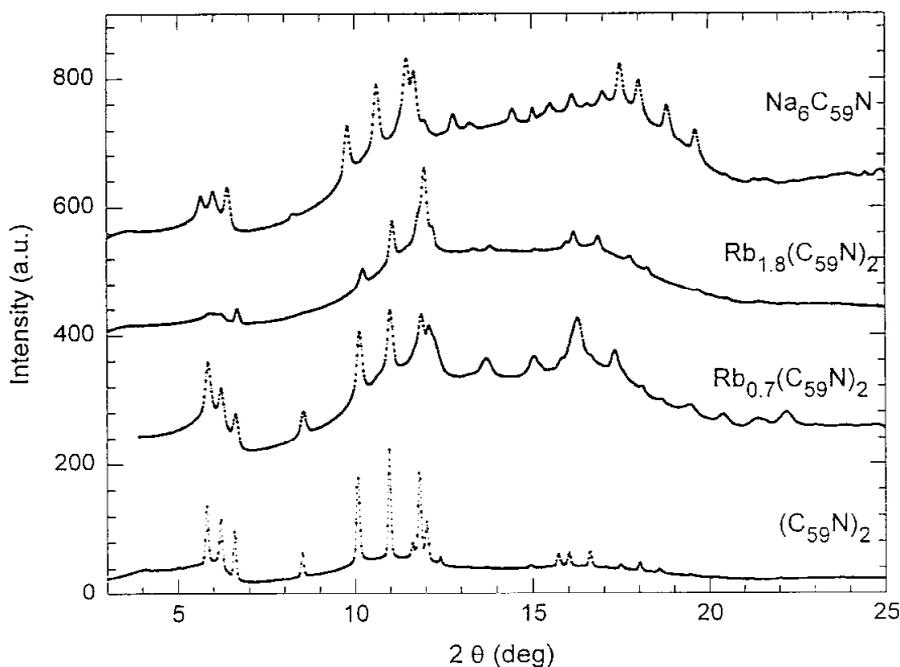


Fig. 1 Synchrotron X-ray diffraction profiles of hexagonal (a) $\text{Na}_6\text{C}_{59}\text{N}$, (b) $\text{Rb}_{0.7}(\text{C}_{59}\text{N})_2$, (c) $\text{Rb}_{1.8}(\text{C}_{59}\text{N})_2$, and (d) $(\text{C}_{59}\text{N})_2$ in the range 3° to 25° at ambient temperature ($\lambda = 0.873 \text{ \AA}$).